Nano-Module 1: Exploring the Properties of Water (H ₂ O)							
Concept	What words cor	ne t	to mind? Properties? Definition?				
Water			•				
Universal solvent							
рН							
Acid							
Base							
Hydrogen Bonding							
Nano-Modul	le 2: Understanding Cher	— nist	try through Molecular Models				
Ionic Bond			Covalent Bond				
metal to nonmetal			nonmetal to				
Electrons			Electrons shared				
T of Moloculo	# -C A tome in Molecu	1.	Everyonles				
Type of Molecule Monatomic	# of Atoms in Molecul	le	Examples				
Diatomic			Of same element: Of different elements:				
Polyatomic			Of same element: Of different elements:				

Name: _____ Group: ____ Date: _____

	T	ype of l	Bond	# of electrons	shared	# of electron pairs]	
=	Single					•		
-	Double		le					
-	Triple						-	
L			None	Madula 2: Ca	rhan Chamist	***	J	
Carbo	n Com	pound	Nano-	ano-Module 3: Carbon Chemistry Properties				
G	Graphit	te						
F	ullerer	ne						
D	Diamon	nd						
				T				
			# protons	# electrons	# neutrons	Radioactive? Yes or No		
		C ₁₂				105 01 110		
		C ₁₃						
		C ₁₄						
Biomolecule Fu		Function		Examples				
Carbo	hydra	ite						
	pids							

Name: _____ Group: ____ Date: _____

Name:	Group:	Date:
Report Template: Introduction: (What do you expect to learn?	What is the purpose of	f this module?)
Materials: (What equipment and material dhardware and software.)	id you need for this	module? Include any special
Procedures: (What steps did you take to acco	omplish this module?)	
Discussion: (Discuss what happened in the nwrong.)	nodules classes. Give	details on anything that went
Conclusion: (Summarize everything you chemicals? How does this compare with your		